

Synthesis of Biaryls Using Nickel-Catalyzed [2+2+2] Cocyclization

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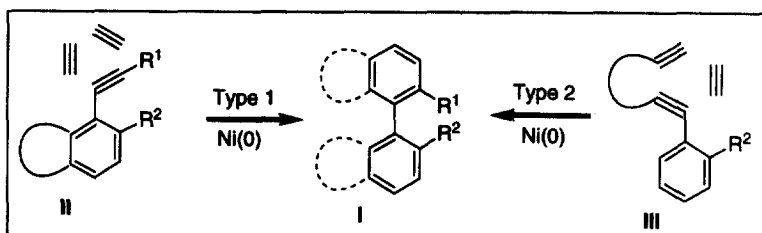
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Abstract: Methods of synthesizing biaryls using nickel-catalyzed [2+2+2] cocyclization were developed. Two ways for the synthesis of biaryl using [2+2+2] cocyclization were investigated: one method is that biaryls synthesized from alkyne having a phenyl group and 2 equivalents of acetylene, and the other method is that those were synthesized from α,ω -diyne having a phenyl group at an α -position and acetylene. The yield is good in each case. © 1999 Elsevier Science Ltd. All rights reserved.

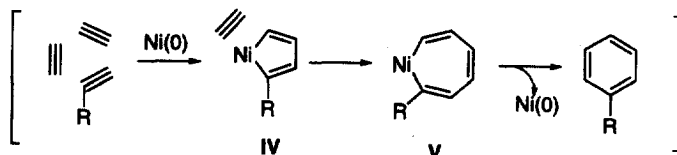
Key Words: Biaryl, [2+2+2] Cocyclization, Cotrimerization, Cycloaddition, Nickel catalyst.

A [2+2+2] cocyclization of alkynes using a transition-metal complex is useful for the construction of the aromatic ring.¹ Recently, the syntheses of the aromatic ring using nickel-promoted² and nickel-catalyzed^{3,4} [2+2+2] cocyclization were reported. Although the reaction mechanism was not clear, the intermediate is thought to be nickelacyclopentadiene **IV** generated from two alkynes by oxidative cyclization, and nickelacycloheptatriene **V**, which was formed by insertion of alkyne into **IV**. These results prompted us to develop a conceptually new methodology for the synthesis of biaryls using a nickel-catalyzed [2+2+2] cocyclization. In general, biaryls are constructed by the attachment of two aryl groups.⁵ However, we planned to construct biaryls by two methods shown in Scheme 1. One method is the synthesis of biaryls from alkynes **II** having an aryl group and 2 equivalents of acetylene (Type 1). The other method is the formation of biaryls from α,ω -dienes **III** having an aryl group at the terminal position and acetylene (Type 2).

Scheme 1. Plan for the Synthesis of Biaryls by Nickel-Catalyzed [2+2+2]Cocyclization

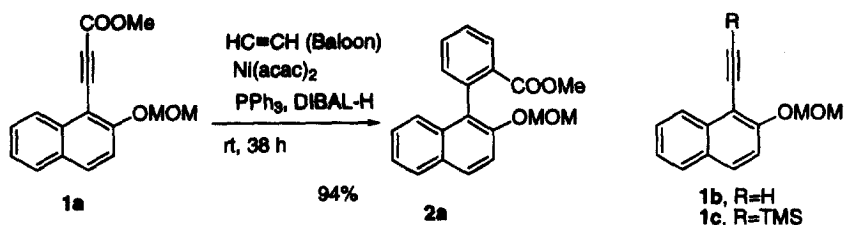


Possible Reaction Course



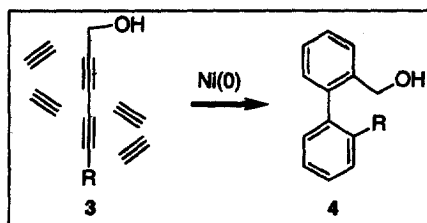
First, we examined the synthesis of biaryls from alkynes having an aryl group and 2 equivalents of acetylene (i.e., Type 1). When a THF solution of naphthalene derivative **1a** having an aryl group at an α -position was stirred at room temperature in the presence of nickel catalyst prepared from $\text{Ni}(\text{acac})_2$ (20 mol %), PPh_3 (80 mol %) and DIBAL-H (40 mol %) under acetylene gas (balloon) for 38 h, biaryl **2a** was obtained in 94% yield. The result was quite interesting because the biaryl skeleton was constructed with an excellent yield from alkyne and gaseous acetylene. On the other hand, the reaction of alkyne **1b** with acetylene under the similar conditions did not afford the desired biaryl but gave complex mixtures along with recovered starting material **1b** in 58%. In the reaction of **1c** under the similar conditions, the starting material **1c** was recovered in 99%. These results suggested that coordination of alkyne to the metal center is important in this cyclization, and that the alkyne moiety in **1a**, having an electron withdrawing substituent, would strongly coordinate to the metal center due to back donation.

Scheme 2. Synthesis of Biaryl from Alkyne +2 Acetylene (Type 1)



It was known that the reaction of α,ω -diyne and propargylic alcohol gave the aromatic compound in good yield because the hydroxy moiety would assist the alkyne to coordinate to the metal center.² Thus, we tried to construct biaryl **4** from 1,3-diyne **3**, having the hydroxymethyl group, and 4 equivalents of acetylene (Scheme 3).

Scheme 3 Plan for the Synthesis of Biaryls from 1,3-Diyne and Acetylene



Scheme 4. Nickel-Catalyzed [2+2+2]Cocyclization Using Diyne and Acetylene

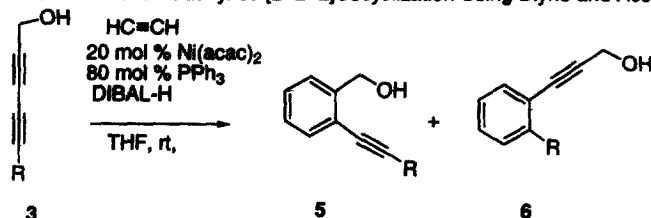
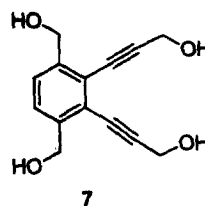


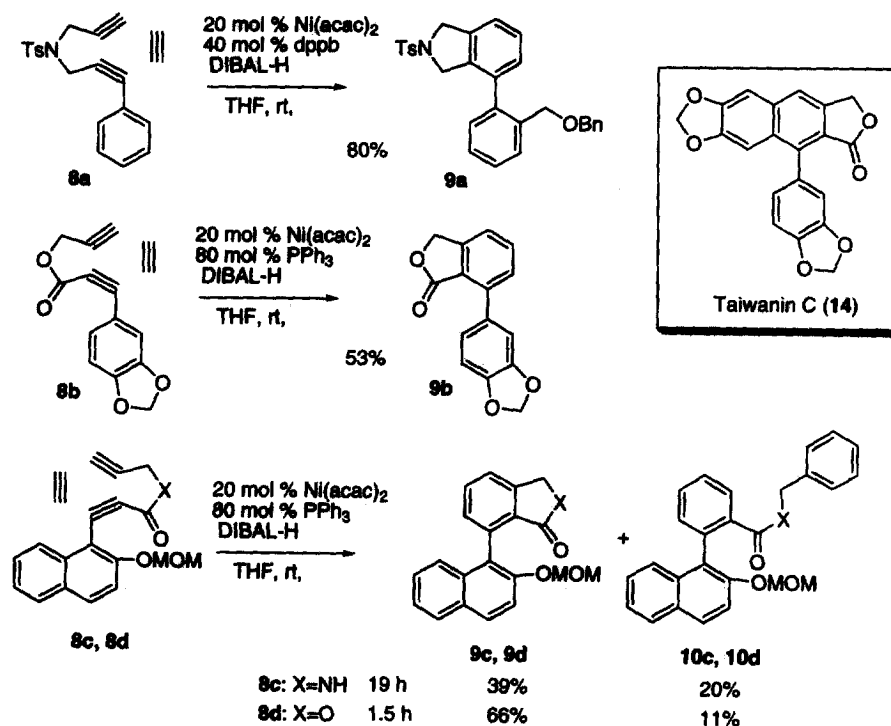
Table 1. Nickel-Catalyzed [2+2+2]Cocyclization of 1,3-Diyne and Acetylene

Run	R		Yield (%)	
			5	6
1	CH ₂ OH	3a	84	—
2	TMS	3b	81	—
3	Bu	3c	80	—
4	Ph	3d	63	—
5	H	3e	29	15
6	COOMe	3f	9	25



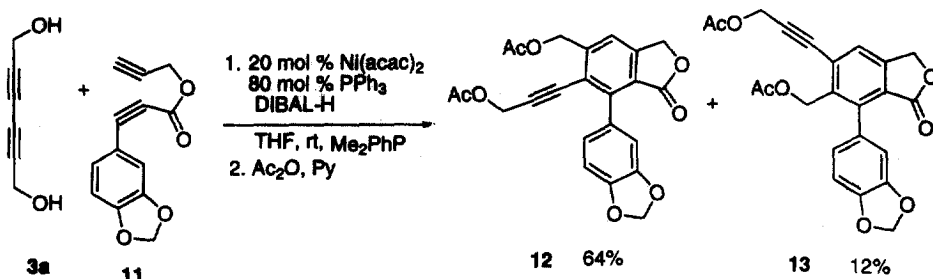
When diyne **3a** was reacted with acetylene in the presence of nickel catalyst, **5a** was obtained in 84% yield (Table 1, run 1), and neither biaryl **4a** nor **7** was obtained. The 1,3-diyne **3b-3e**, having various substituents, were used for this reaction, however, the desired biaryls **4** were not formed (Table 1), and in each case, the main product was benzyl alcohol **5**. When the alkyne **3f**, having a carbomethoxy group, was used, the main product was changed from **5** to **6** (run 6), and the result is consistent with that of the cyclization of **1a**. Although the construction of biaryl was unsuccessful in this system, the high reactivity of di-substituted alkyne **3** on [2+2+2] cocyclization was quite interesting.⁶

Subsequently, we examined the construction of biaryls from α,ω -diyne and alkyne (i.e. Type 2). When a THF solution of diyne **8a** having a phenyl group was stirred under acetylene gas in the presence of nickel catalyst derived from Ni(acac)₂, dppb, and DIBAL-H at room temperature, the desired biaryl **9a** was obtained in 80% yield.

Scheme 5. Synthesis of Biaryls from α,ω -Diyne and Alkyne

In the reaction of diyne **8b**, having a (methylenedioxy)phenyl group, with acetylene, the desired biaryl **9b** was obtained in 53% yield. The alkynes, **8c** and **8d**, having a naphthalene ring, gave biaryls **9c** and **9d** in 39% and 66% yields along with biaryls **10c** and **10d** in 20% and 11% yields, respectively. Since the reactivity of diyne **3** was quite interesting as a functionalized di-substituted alkyne, the reaction of diyne **3a** with **11** was also investigated. The reaction resulted in biaryls **12** and **13** being obtained in 64% and 12% yields, respectively. The biaryl **12** should be a good intermediate for the synthesis of natural products containing biaryl skeletons such as Taiwanin C (**14**). The results indicate that various biaryls having functional groups could be obtained from α,ω -diynes and alkynes, and that di-substituted alkyne **3a** can be used for the synthesis of biaryl as well as acetylene and DMAD.

Scheme 6



In conclusion, biaryls were constructed from alkyne and two equivalents of acetylene or from α,ω -diyne and alkyne using nickel-catalyzed [2+2+2] cycloaddition in good to excellent yields. In this reaction, 1,3-diyne **3a** could be used as di-substituted alkyne. Further studies on the application of the synthesis of natural products as well as on the asymmetric synthesis of biaryls are in progress.

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